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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=173 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
$R$ factor $=0.045$
$w R$ factor $=0.114$
Data-to-parameter ratio $=17.7$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## rac-(Diisopropylamino)(1-indenyl)(isopropylamino)borane

In the racemic title compound, $\mathrm{C}_{18} \mathrm{H}_{29} \mathrm{BN}_{2}$, the B and the N atoms are virtually trigonal-planar coordinated and the presence of $\mathrm{B}=\mathrm{N}$ partial double bonding is confirmed by the $\mathrm{B}-\mathrm{N}$ bond lengths of 1.4154 (18) and 1.4178 (18) $\AA$.

## Comment

Ligand precursors related to the title compound rac-(diisopropylamino)(1-indenyl)(isopropylamino)borane, (I), have been previously employed for the synthesis of boronbridged constrained-geometry complexes. These complexes exhibit activity for the polymerization of ethene and higher $\alpha$ olefins when activated with MAO (methylaluminoxane) (Braunschweig et al., 2000, 2004). In comparative studies, a strong influence of the ligand substitution patterns in corresponding complexes has been demonstrated on both catalyst activity and polymerization characteristics (Stevens 1994), justifying the synthesis of a series of boron-bridged analogues.

(I)

The structure of (I) (Fig. 1) in the solid state largely resembles that of its corresponding $N$-phenyl-substituted counterpart $\left(\eta^{1}-\mathrm{C}_{9} \mathrm{H}_{7}\right) \mathrm{B}\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right) \mathrm{N}(\mathrm{H}) \mathrm{Ph}$ (Braunschweig et al., 2003). Compound (I) adopts $C_{1}$ symmetry in the crystal structure. The crystal structure confirms the constitution assumed by NMR spectroscopy, with boron being attached to the $s p^{3}$-hybridized C atom of the indenyl system. The B1 and N 2 centres adopt trigonal-planar geometries, with the central atoms lying only 0.015 (B1) and $0.025 \AA$ (N2) out of the planes of their respective substituents. These two trigonal planes are virtually coplanar with torsion angles of 176.35 (12) (C16$\mathrm{N} 2-\mathrm{B} 1-\mathrm{N} 1)$ and $0.2(2)^{\circ}(\mathrm{C} 13-\mathrm{N} 2-\mathrm{B} 1-\mathrm{N} 1)$. The $\mathrm{B} 1-\mathrm{N} 1$ and $\mathrm{B} 1-\mathrm{N} 2$ bond lengths of 1.4154 (18) and 1.4178 (18) $\AA$, respectively, are equivalent within experimental error, indicating equal $\pi$-contributions from both amino groups. The N2-

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bound isopropyl group syn to the indenyl moiety (which is coplanar to within $0.007 \AA$ ) is oriented such that the methine proton is directed towards the middle of the five-membered ring, a conformation also seen in the corresponding compound $\left.\left(\mathrm{C}_{9} \mathrm{H}_{7}\right) \mathrm{B}\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right) \mathrm{N}(\mathrm{H}) \mathrm{Ph}\right)$ with $\mathrm{H}-\pi$ distances to the centres of the $\mathrm{C} 2=\mathrm{C} 3$ olefinic and $\mathrm{C} 1=\mathrm{C} 9$ aromatic bonds of 2.446 and $2.509 \AA$, respectively. No short intermolecular contacts were observed.

## Experimental

The title compound was prepared by a modified literature procedure (Braunschweig et al., 2003) by sequential reaction of ${ }^{i} \mathrm{Pr}_{2} \mathrm{NBCl}_{2}$ $(2.16 \mathrm{~g}, 11.9 \mathrm{mmol})$ with lithium indenide ( $1.45 \mathrm{~g}, 11.9 \mathrm{mmol}$ ) and lithium isopropylamide $(0.77 \mathrm{~g}, 11.9 \mathrm{mmol})$ in a hexane-toluene mixture. Recrystallization from hexane at 243 K afforded (I) exclusively as the 1 -indenyl isomer in $35 \%$ yield in the form of pale-yellow crystals suitable for single-crystal X-ray diffraction.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{18} \mathrm{H}_{29} \mathrm{BN}_{2} \\
& M_{r}=284.24 \\
& \text { Monoclinic, } P 2_{1} / n \\
& a=8.1029(8) \AA \\
& b=14.0352(14) \AA \\
& c=15.6679(15) \AA \\
& \beta=100.503(2)^{\circ} \\
& V=1752.0(3) \AA^{\circ} \\
& Z=4
\end{aligned}
$$

$D_{x}=1.078 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
Cell parameters from 4341
$\quad$ reflections
$\theta=2.6-26.1^{\circ}$
$\mu=0.06 \mathrm{~mm}^{-1}$
$T=173(2) \mathrm{K}$
Tablet, yellow
$0.25 \times 0.20 \times 0.09 \mathrm{~mm}$

Data collection

| Bruker SMART CCD area-detector | 3471 independent reflections |
| :--- | :--- |
| $\quad$ diffractometer | 2751 reflections with $I>2 \sigma(I)$ |
| $\varphi$ and $\omega$ scans | $R_{\text {int }}=0.032$ |
| Absorption correction: multi-scan | $\theta_{\max }=26.1^{\circ}$ |
| $\quad(S A D A B S ;$ Sheldrick, 2002) | $h=-10 \rightarrow 10$ |
| $T_{\min }=0.972, T_{\max }=0.99$ | $k=-17 \rightarrow 17$ |
| 16287 measured reflections | $l=-19 \rightarrow 19$ |
| Refinement |  |
| Refinement on $F^{2}$ | $w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0505 P)^{2}\right.$ |
| $R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.045$ | $+0.473 P]$ |
| $w R\left(F^{2}\right)=0.114$ | where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$ |
| $S=1.02$ | $(\Delta / \sigma)_{\max }=0.001$ |
| 3471 reflections | $\Delta \rho_{\max }=0.23 \mathrm{e} \AA^{-3}$ |
| 196 parameters | $\Delta \rho_{\min }=-0.19 \mathrm{e}^{-3}$ |
| H-atom parameters constrained |  |

H atoms were visible in difference Fourier maps and were subsequently placed at idealized positions and treated as riding atoms, with $\mathrm{C}-\mathrm{H}=0.98\left(\mathrm{CH}_{3}\right), 0.95$ (aromatic CH$)$ and $1.00 \AA$ (aliphatic CH ), and $\mathrm{N}-\mathrm{H}=0.88 \AA$. $U_{\text {iso }}$ values for primary H atoms were fixed at 1.5 times and tertiary H atoms at 1.2 times $U_{\mathrm{eq}}(\mathrm{C})$.


Figure 1
A view of (I), with displacement ellipsoids drawn at the $30 \%$ probability level.

Data collection: SMART-NT (Bruker, 2001); cell refinement: SAINT-Plus-NT (Bruker, 2001); data reduction: SAINT-Plus-NT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: XP in SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: $X P$ in $S H E L X T L$.

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